14.877(4) Å, z = 2, and space group *Pnmm*. K₃TiTa₇O₂₁ crystallizes in the hexagonal system with unitcell dimensions a = 9.095(3), c = 12.063(4) Å, z = 2 and space group $P6_3/mcm$. The structures were solved by conventional Patterson and Fourier techniques, the former using 283 unique reflections ($I \ge$ $3\sigma(I)$) was refined by full-matrix least-squares techniques to an R of 0.0551 ($R_w = 0.0496$), the latter using 343 unique reflections ($I \ge 3\sigma(I)$) was refined using the same technique to an R of 0.0354 ($R_w =$ 0.0279). Intensity data were measured using counter methods and the " ω " scan technique. KTi₃TaO₉ is isostructural with KTi₃NbO₉ whose structure was previously derived from powder data, and K₃TiTa₇O₂₁ adopts the same structure as found for the nonstoichiometric phase K_{5.5}Ta_{15.7}O₄₂ in this laboratory. Both materials were obtained as single crystals by the use of a flux in their preparation. The distribution of titanium and tantalum over the octahedral sites is discussed relative to previously determined structures.

Kinetics and Mechanistic Aspects of the Reduction of Chromium Dioxide. R. SAEZ-PUCHE AND M. A. ALARIO-FRANCO,* Laboratoire de Cristallographie, CNRS 166 X, 38042 Grenoble Cédex, France. Thermogravimetric experiments on very small, single-crystalline, acicular particles of CrO_2 , of homogeneous size and shape, indicate that the reduction of CrO_2 by hydrogen, to produce CrOOH, can be explained by means of a unidimensional diffusion mechanism. Hydrogen diffuses along the empty tunnels that exist parallel to the *c*-axis in the rutile-like unit cell. The corresponding diffusion equation is: $D = 10^{-6} (-19.3 \pm 2.3/RT) \text{ cm}^2/\text{sec}$, and is valid for 490 K < T < 519 K and $30 < P_{H_2} < 110$ Torr.

Rare Earth Transition Metal Sulfides, $LnMS_3$ (Ln = Rare Earth; M = Transition Metal). T. MURUGESAN, S. RAMESH, J. GOPALAKRISHNAN, AND C. N. R. RAO,* Solid State and Structual Chemistry Unit, Indian Institute of Science, Bangalore, 560012, India. Ternary rare earth-transition metal sulfides $LnMS_3$ with Ln = La, Nd, Gd, and M = V, Cr as well as Ln = La and M = Mn, Fe, Co, Ni have been prepared and characterized. The vanadium and chromium sulfides crystallize in a monoclinic layer structure isotypic with $LaCrS_3$ while the other $LnMS_3$ sulfides crystallize in a hexagonal structure. Chemical shifts of the metal K-absorption edge and XPS binding energies of core levels indicate that the transition metal is trivalent in the V and Cr sulfides, while it is divalent in the Mn, Fe, Co, and Ni sulfides. Electrical and magnetic properties of the sulfides are discussed in terms of their structure and electronic configurations of the transition metal ions.

Kationenverteilungen in einigen Zinnspinell-Mischkristallen. G. VON BECKH, P. ZEGREANU, AND M. TROMEL,* Institut fur Anorganische Chemie der Johann Wolfgang Goethe Universitat, Niederurseler Hang, D-6000 Frankfurt a.M. 50, Bundesrepublik Deutschland. In inverse tin spinels (Mg, M)₂SnO₄ (with M = Zn, Co, or Ni), the distribution of bivalent metal atoms was determined from samples which had been quenched from 1473 to 1073 K or cooled slowly. In Mg₂SnO₄, zinc preferentially occupies tetrahedral sites with ΔG° values of about 18 kJ/g-atom. Cobalt and magnesium are equally distributed over octahedral and tetrahedral sites, whereas nickel occupies octahedral sites only. Lattice constants of (Mg, Zn)₂SnO₄ solid solutions deviate largely from Vegard's law, but depend linearly on the zinc concentrations in tetrahedral and octahedral sites according to $a = c_0 + c_1 [Zn(tetr.)] + c_2 [Zn(oct.)]$. Zinc in tetrahedral sites causes a lattice contraction; in octahedral sites, however, a lattice expansion.

Deuteron Magnetic Resonance and Hydrogen Bond Network of Ammonium Trihydrogen Selenite. I. S. VINOGRADOVA, L. V. Kirensky Institute of Physics, Academy of Sciences USSR, Siberian Branch, Krasnoyarsk, 660036, USSR. The DMR spectra of single-crystal $ND_4D_3(SeO_3)_2$ have been studied. The principal values and the direction cosines of field-gradient tensor of deuterons located on three nonequivalent O. . .O hydrogen bonds have been determined. The lengths of hydrogen bonds have been calculated from eQq/h values; the deuterons have been located on hydrogen bonds. The comparison with the DMR data of isomorphous compound $RbD_3(SeO_3)_2$ is made, and influence of N– H. . .O hydrogen bonds on the structural parameters of O. . .O hydrogen bonds is discussed.

Les Differentes Formes Cristallines de Y_3ReO_{θ} . Relations avec la Structure Fluorine. G. BAUD, J. -P. BESSE, R. CHEVALIER,* AND M. GASPERIN, Equipe de Cristallographie et Physico-Chimie des Matériaux, associée au CNRS (ERA 897), Université de Clermont II, B.P. 45, 63170 Aubiere, France. The monoclinic $Y_3ReO_{\theta}\beta$ phase is synthesized at 40 kbar and 900°C. The crystal structure is solved to